Chemical Product Design

E. L. Cussler

University of Minnesota

G. D. Moggridge

University of Cambridge



PUBLISHED BY THE PRESS SYNDICATE OF THE UNIVERSITY OF CAMBRIDGE The Pitt Building, Trumpington Street, Cambridge, United Kingdom

CAMBRIDGE UNIVERSITY PRESS

The Edinburgh Building, Cambridge CB2 2RU, UK 40 West 20th Street, New York, NY 10011-4211, USA 10 Stamford Road, Oakleigh, VIC 3166, Australia Ruiz de Alarcón 13, 28014 Madrid, Spain Dock House, The Waterfront, Cape Town 8001, South Africa http://www.cambridge.org

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First published 2001

Printed in the United States of America

Typefaces Times Ten Roman 10/13 pt. and Gill Sans System LATEX $2_{\mathcal{E}}$ [TB]

A catalog record for this book is available from the British Library.

Library of Congress Cataloging-in-Publication Data Cussler, E. L.

Chemical product design / E. L. Cussler, G. D. Moggridge.

p. cm. – (Cambridge series in chemical engineering)

ISBN 0-521-79183-9 - ISBN 0-521-79633-4 (pb)

1. Chemical industry. I. Moggridge, G. D. II. Title. III. Series. TP149 .C85 2001 660'.068'5 – dc21 00-063069

ISBN 0 521 79183 9 hardback

ISBN 0 521 79633 4 paperback

Contents

| | Pref | face | page xi |
|---|------|--|---------|
| | Note | ntion | XV |
| ı | An | Introduction to Chemical Product Design | 1 |
| | 1.1 | What Is Chemical Product Design? | 1 |
| | 1.2 | Why Chemical Product Design Is Important | 3 |
| | | Changes in the Chemical Industry | 3 |
| | | Changes in Employment | 5 |
| | 1.3 | Changes in Corporate Culture | 6 |
| | | Corporate Organization | 7 |
| | | Corporate Strategy | 8 |
| | 1.4 | The Product Design Procedure | 8 |
| | | How the Procedure Organizes this Book | 9 |
| | | Limitations of the Procedure | 9 |
| | 1.5 | Conclusions | 11 |
| 2 | Nee | eds | 13 |
| | 2.1 | Customer Needs | 13 |
| | | Interviewing Customers | 13 |
| | | Interpreting Customer Needs | 15 |
| | | Example 2.1–1. Better Thermopane Windows | 16 |
| | | Example 2.1–2. Alternative Fluids for Deicing | |
| | | Airplanes | 18 |
| | | Example 2.1–3. "Smart" Labels | 20 |
| | 2.2 | Consumer Products | 22 |
| | | Consumer Assessments | 23 |
| | | Consumer versus Instrumental Assessments | 24 |
| | | Example 2.2–1. Tasty Chocolate | 25 |
| | | Example 2.2–2. The Consumer Attribute "Viscosity" | 26 |
| | 2.3 | Converting Needs to Specifications | 27 |
| | | Example 2.3–1. Muffler Design | 28 |
| | | Example 2.3–2. Water Purification for the Traveler | 29 |
| | | Example 2.3–3. Preventing Explosions in High-Performance | |
| | | Batteries | 30 |

| | 2.4 | Revising Product Specifications | 33 |
|---|------|---|----|
| | | Example 2.4–1. Deicing Winter Roads | 34 |
| | | Example 2.4–2. Scrubbing Nitrogen from Natural Gas | 38 |
| | 2.5 | Conclusions and the First Gate | 41 |
| 3 | ldea | as | 43 |
| | 3.1 | Human Sources of Ideas | 44 |
| | | Sources of Ideas | 44 |
| | | Collecting the Ideas | 45 |
| | | Problem Solving Styles | 46 |
| | | Examples of Unsorted Ideas | 48 |
| | 3.2 | Chemical Sources of Ideas | 49 |
| | | Natural Product Screening | 52 |
| | | Random Molecular Assembly | 54 |
| | | Combinatorial Chemistry | 55 |
| | | Example 3.2–1. Fuel Cell Catalysis | 57 |
| | 3.3 | Sorting the Ideas | 57 |
| | | Getting Started | 58 |
| | | "The Material Will Tell You" | 58 |
| | | Example 3.3–1. Adhesives for Wet Metal | 60 |
| | | Example 3.3–2. Reusable Laundry Detergents | 61 |
| | 2.4 | Example 3.3–3. Pollution Preventing Ink | 63 |
| | 3.4 | Screening the Ideas | 64 |
| | | Strategies for Idea Screening | 65 |
| | | Improving the Idea Screening Process | 66 |
| | | Example 3.4–1. Home Oxygen Supply | 68 |
| | 2.5 | Example 3.4–2. High-Level Radioactive Waste | 69 |
| | 3.5 | Conclusions and the Second Gate | 73 |
| 4 | Sele | ection | 75 |
| | 4.1 | Selection Using Thermodynamics | 76 |
| | | Ingredient Substitutions | 76 |
| | | Substitutions in Consumer Products | 77 |
| | | Ingredient Improvements | 79 |
| | | Example 4.1–1. A Better Skin Lotion | 80 |
| | | Example 4.1–2. A Pollution Preventing Ink | 81 |
| | | Example 4.1–3. Antibiotic Purification | 81 |
| | 4.2 | Selection Using Kinetics | 82 |
| | | Chemical Kinetics | 82 |
| | | Heat and Mass Transfer Coefficients | 84 |
| | | Example 4.2–1. A Device that Allows Wine to Breathe | 85 |
| | 4.2 | Example 4.2–2. A Perfect Coffee Cup | 87 |
| | 4.3 | Less Objective Criteria | 90 |
| | | When to Make Subjective Judgments | 91 |
| | | How to Make Subjective Judgments | 92 |
| | | Why We Use Selection Matrices | 93 |
| | | Example 4.3–1. Monarchy Substitution | 94 |
| | | Example 4.3–2. The Home Ventilator | 94 |

Contents ix

| | 4.4 | Risk in Product Selection | 102 |
|---|-----|---|------------|
| | | Risk Assessment | 103 |
| | | Risk Management | 105 |
| | | Example 4.4–1. Power for Isolated Homes | 107 |
| | | Example 4.4–2. Taking Water out of Milk at the Farm | 109 |
| | 4.5 | Conclusions and the Third Gate | 114 |
| 5 | Pro | duct Manufacture | 116 |
| | 5.1 | Intellectual Property | 117 |
| | | Patents and Trade Secrets | 118 |
| | | What Can Be Patented | 120 |
| | | Requirements for Patents | 120 |
| | | Example 5.1–1. The Invention of the Windsurfer | 123 |
| | 5.2 | Supplying Missing Information | 123 |
| | | Reaction Path Strategies | 124 |
| | | Example 5.2–1. Synthesis of the Tranquilizer, Phenoglycodol | 125 |
| | | Example 5.2–2. Sterically Hindered Amines for CO ₂ Removal | |
| | | from Gases | 125 |
| | | Example 5.2–3. Silver Bullets for Zebra Mussels | 127 |
| | 5.3 | Final Specifications | 128 |
| | | Product Structure | 129 |
| | | Central Product Attributes | 130 |
| | | Chemical Triggers | 130 |
| | | Example 5.3–1. Freon-Free Foam | 131 |
| | | Example 5.3–2. Better Blood Oxygenators | 134 |
| | 5.4 | Microstructured Products | 137 |
| | | Thermodynamics | 139 |
| | | Colloid Stability | 141 |
| | | Rheology and Mixing | 144 |
| | | Example 5.4–1. Destabilizing Latex Paint | 146 |
| | ~ ~ | Example 5.4–2. Making More Ice Cream | 147 |
| | 5.5 | Device Manufacture | 148 |
| | | Thermodynamics | 148 |
| | | Enzyme Kinetics | 150 |
| | | Example 5.5–1. An Electrode for Measuring Dodecyl Sulfate | 151 |
| | 5.6 | Example 5.5–2. Designing an Osmotic Pump Conclusions | 152 154 |
| | | | 134 |
| 6 | - | cialty Chemical Manufacture | 156 |
| | 6.1 | First Steps Toward Production | 157 |
| | | Extending Laboratory Results | 158 |
| | | Reaction Engineering | 160 |
| | | Example 6.1–1. Penicillin Modification | 160 |
| | | Example 6.1–2. Etching a Photoresist | 161 |
| | 6.2 | Separations | 162 |
| | | Heuristics for Separations | 163 |
| | | The Most Useful Separations | 166 |
| | | Example 6.2–1. Penicillin Purification | 176 |

x Contents

| | 6.3 | Specialty Scale-Up | 177 |
|---|------|---|-----|
| | | Reactor Scale-Up | 178 |
| | | Separation Scale-Up | 181 |
| | | Example 6.3–1. Reacting Suspended Steroids | 185 |
| | | Example 6.3–2. Scaling Up a Lincomycin Adsorption | 185 |
| | 6.4 | Conclusions | 187 |
| 7 | Eco | nomic Concerns | 189 |
| | 7.1 | Product versus Process Design | 190 |
| | | Commodity Products | 191 |
| | | Specialty Products | 192 |
| | 7.2 | Process Economics | 193 |
| | | A Hierarchy of Process Design | 193 |
| | | Economic Potential | 196 |
| | | Capital Requirements | 198 |
| | 7.3 | Economics for Products | 200 |
| | | Cash Flow Without the Time Value of Money | 202 |
| | | Cash Flow Including the Time Value of Money | 204 |
| | | Time To Market | 206 |
| | | Example 7.3–1. The Economics of Scottish Mussel Farming | 207 |
| | 7.4 | Conclusions and the Fourth Gate | 208 |
| | Prol | plems | 211 |
| | Inde | x | 227 |
| | Prod | lucts Index | 229 |

An Introduction to Chemical Product Design

This is a book about the design of chemical products. In our definition of chemical products, we include three categories. First, there are new specialty chemicals that provide a specific benefit. Pharmaceuticals are the obvious example. Second, there are products whose microstructure, rather than molecular structure, creates value. Paint and ice cream are examples. The third category of chemical products is devices that effect chemical change. An example is the blood oxygenator used in open-heart surgery.

The nature of chemical product design is described in Section 1.1. Product design emphasizes decisions made before those of chemical process design, a more familiar topic. Chemical product design is a response to major changes in the chemical industry that have occurred in recent decades. These changes, described in Sections 1.2 and 1.3, involve a split in the industry between manufacturers of commodity chemicals and developers of specialty chemicals and other chemical products. The former are best served by process design, and the latter by product design.

The fourth section of this chapter outlines the product design procedure that we will use in the remainder of the book. This procedure is a simplification of those already used in business development. Such a simplification clarifies the basic sequence of ideas involved. Moreover, the simple procedure allows us to consider in considerable detail the technical questions implied in specific products. This technical approach is suitable for those with formal training in engineering and chemistry, and may also be challenging for those whose training is largely in business.

1.1 What Is Chemical Product Design?

Imagine four chemically based products: an amine for scrubbing acid gases, a pollution-preventing ink, an electrode separator for high-performance batteries, and a ventilator for a well-insulated house.

These four products may seem to have nothing in common. The amine is chemically well defined: a single chemical species capable of selectively reacting with

sulphur oxides. The ink is a chemical mixture, which includes both a pigment and a polydisperse polymer "resin." The electrode separator should provide a safeguard against explosion if the battery accidentally shorts out. The ventilator both provides fresh air and recovers the energy carefully conserved by insulating the house in the first place.

What these products do have in common is the procedure by which they may be designed. In each case, we begin by defining what we need. Next, we think of ideas to meet this need. We then select the best of these ideas. Finally, we decide what the product should look like and how it should be manufactured.

We define chemical product design as this entire procedure. At the start of the procedure, when we are deciding what the product should do, we expect major input from both marketing and research, as well as from engineering. By the end of the procedure, when we are focused on the manufacturing process, we expect a reduced role for marketing, and a major effort from engineering. However, we believe that the entire effort is best viewed as a whole, carried out by integrated teams drawn from marketing, research, and engineering.

We can see how product design develops by considering three of the products already mentioned in somewhat more detail. For example, for the pollution-preventing ink, our original need may be to reduce emissions of volatile solvents in the ink by 90%. Our ideas to meet this need include reformulating the polymer resin in the ink in two different ways. First, by using a polydisperse resin of broader molecular weight distribution, we can eliminate the need for volatile solvents in the ink itself. Thus there will be no emissions during printing. Second, by adding pendant carboxylic acid groups to the resin, we can make the resin not only an effective component of the ink but also an emulsifying agent in dilute base. If we wash the presses with dilute base, we can clean them without volatile solvents and without solvent-soaked shop rags. The manufacture of the new ink will be similar to that used for the existing ink.

Consider the amine for scrubbing acid gases. Current acid gas treating often uses aqueous solutions of amines, such as monoethanol amine. After these solutions absorb acid gases such as carbon dioxide and sulphur oxides, they are regenerated by heating. Though this heating gives an efficient regeneration, it can be expensive. The need is for amines that can be more easily regenerated. Our idea is to effect the regeneration with changes in pressure. We would absorb the acid gases at high pressure and regenerate the amines at low pressure, where the acid gases just bubble out of solution. In order to achieve this end we have little idea how to proceed, so we are forced to synthesize small amounts of a large number of sterically hindered amines. We will test all candidates to find the best ones. We will then manufacture the winners. Like many high value-added chemicals, these will be custom syntheses, made in batches in equipment used for a wide variety of products. This obviates the need for intensive process design in many of the chemical products that we are considering.

A third example of product design is house ventilation. Well-insulated houses are energy efficient, costing little to heat, but they may exchange air at less than one tenth of the recommended rate. To get more fresh air, we can open a window, but

this sacrifices our efforts at good insulation. The need is for a fresh air exchanger that captures the heat and humidity of our snug house, but exhausts stale air, with smells and carbon dioxide. Our idea is for an exchanger for both heat and water vapour for this energy efficient house. We can manufacture this in the same way as other low cost, cross flow heat exchangers. In this example, our product is a device – not a chemical – that increases health and comfort in the house.

The designs of the ink, the acid gas absorbant, and the home ventilator are examples of the subject of this book. This subject is different from chemical process design. In process design, we normally begin by knowing what the product is, and what it is for. Most commonly, it is a commodity chemical of carefully defined purity; ethylene and benzene are good examples. This material will be sold into the existing market for such a commodity, so we will know the price we can expect. The focus of our process design will be efficient manufacture. We will usually use a continuous process, depending on optimized, dedicated equipment, which has been thoroughly energy integrated. This type of careful process design is essential in order to compete successfully in the commodity chemical business.

The chemical products discussed in this book are completely different. Their promise stems less from their efficient manufacture, and more from their special functions. They will usually be made in batch, in generic equipment; or will themselves be small pieces of equipment. Process efficiency may be less important than speed to reach the marketplace. Energy integration may be of secondary value. Indeed, most of product design may occur before manufacture is even an issue.

We believe that product design merits increased emphasis because of major changes in the chemical industry. We do not argue that the chemical engineer's concern with process design should disappear. However, we do assert that the topics we study should reflect the chemical industry of today. How this has developed is outlined in the next section.

1.2 Why Chemical Product Design Is Important

Chemical product design has become more important because of major changes in the chemical industry. To understand these changes, we will review the history of the industry, using as an example the development of synthetic textile fibers. We also need to examine how these changes have affected employment.

CHANGES IN THE CHEMICAL INDUSTRY

From 1950 to 1970 the chemical industry produced ever increasing amounts of synthetic textile fibers, as shown in Table 1.2–1. Over these decades, while the production of natural fibers was about constant, the production of synthetics grew 20% per year. This growth was like that of the software industry today; Du Pont can be seen as the Microsoft of the 1950s. This was a golden age for chemicals.

However, from 1970 to 1990, synthetic textile fibers grew by less than 5% per year, about the same as the growth of the world population. From 1970 to 1990, the

TABLE 1.2-I Growth of Textile Fibers (106 lbs/year)

| Fiber | 1948 | 1969 | 1989 |
|--------------|------|------|------|
| Cotton, wool | 4353 | 4285 | 4794 |
| Synthetics | 92 | 3480 | 8612 |

Note: From 1950 to 1970, synthetic fibers grew about 20% per year. Since then, their growth is 5% per year (source: Spitz (1988); U.S. Department of Commerce).

industry stayed profitable by using larger and larger facilities. Bigger profits came from consolidating production into bigger plants, designed for greater efficiency in making one particular product. Interest in computer-optimized design was a consequence of this consolidation. Such optimization meant small producers were forced out. For example, the number of companies making vinyl chloride in the USA shrank from twelve in 1964 to only six in 1972 (Spitz, 1988).

More recently, the industry has required new strategies to stay profitable. These strategies often centered on restructuring, which was three times more likely to affect engineers than the general working population. Whether called restructuring, downsizing, rightsizing, or rationalization, the strategy meant many midcareer engineers were suddenly looking for new jobs. The Engineering Workforce Commission in the USA now feels that engineers will average seven different jobs per career, a dramatic change from two per career in the recent past (Ellis, 1994). Middle management, that traditional goal of bright but not brilliant students, is no longer a safe haven. While starting salaries remain high, the envy of other technical professions, these salaries have not increased faster than average wage inflation in 30 years. In this environment, professional organizations such as the American Institute of Chemical Engineers now provide more help in job transitions and financial planning. Such organizations can no longer behave only as nineteenth-century-style learned societies.

Having exhausted optimization and restructuring as ways to stay profitable, chemical companies now have three remaining options. First, they can leave the chemical business. This option seems reasonable to a surprising number, including many petrochemical businesses. Second, chemical companies can focus exclusively on commodities. This seems a preferred strategy for some private companies, who may be better able to handle the ebb and flow of the profits from a commodity business. It implies a ruthless minimization of research and a concentration on in-house efficiency.

The third strategy open to these chemical companies is to focus their growth on specialty chemicals. Such chemicals, produced in much smaller volumes than commodities, typically have much higher added value as well. These higher added values mean that more research and higher profits are possible. Not surprisingly, many chemical companies are turning their focus to specialty chemicals.

Interestingly, this new focus has not changed the skills that companies demand from chemists and chemical engineers, though it has changed the jobs that they do.

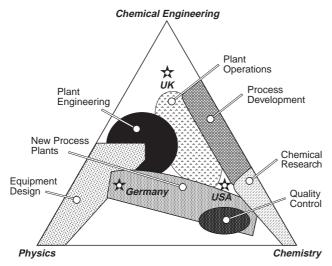


Figure 1.2–1. Skills Learned by Chemical Engineers. Skills traditionally learned by chemical engineers are a blend of chemical engineering, chemistry, and physics (including mechanics). These skills do not need major changes to be valuable for products.

The various subjects that chemical engineers learn can be positioned on the triangular diagram in Figure 1.2–1. The three corners of this plot represent training in physical sciences, in the chemical sciences, and in chemical engineering subjects. Different jobs use these three elements in different proportions, as shown in the figure. There is no surprise in this: plant engineering will demand a greater knowledge of mechanics and a smaller background in chemistry than in research and development. Figure 1.2–1 also suggests national averages. British chemical engineers have more chemical engineering and less chemistry than their counterparts elsewhere. Please do not take this diagram too literally; use it instead as a catalyst for thought. We would maintain that the basic skills needed by chemical engineers have always been diverse and have not altered dramatically.

CHANGES IN EMPLOYMENT

Although changes in the chemical industry may not have changed the skills needed, the focus of chemical companies on specialties has had a major impact on the jobs that chemists and chemical engineers do. To examine this impact, we compare the jobs taken by recent graduates with those taken by graduates 25 years ago. Our data for this are fragmentary, taken from records of graduates from the Universities of Cambridge and Minnesota. They are probably biased toward large corporations, about whom our university placement offices have better records.

The available data suggest major changes, as shown in Figure 1.2–2. In 1975, three quarters of chemical engineering graduates went to work in the commodity chemicals business. The small number who did not were split between work

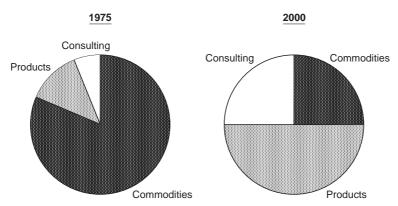


Figure 1.2–2. Changes in Employment. The dominance of commodity chemicals in the past has been eroded by the newer emphasis on products, including specialty chemicals.

on products, either design or development; and work in other areas, which for convenience we have labeled "consulting." This category includes those working directly for consulting firms as well as those carrying out specific tasks such as environmental impact statements.

More recently, the distribution of jobs has become completely different. The largest group of chemical engineering graduates, in Minnesota's case more than half, now work primarily on products. This includes students who work on materials, on coatings, on adhesives, and on specialty chemicals. The number who work in commodity chemicals has dropped so that it now is less than a quarter of new graduates. The number who work in consulting has risen dramatically, as commodity chemical businesses outsource many of the functions that they used to do in house. In one case, a commodity chemical company has taken its process engineering group from 1500 persons to fewer than 50. This is not a business cycle; this is a change in the way in which that company expects to do business; they will buy the process engineering they need from consultants. This is why the number of people involved in consulting has risen.

The emergence of products as a focus for chemical engineers implies changes in what chemical engineers do. In the past, we chemical engineers could limit our thinking to reaction engineering and unit operations, waiting for the marketing division to tell us what chemicals needed to be made, and in what amounts. Such intellectual isolation is no longer possible.

1.3 Changes in Corporate Culture

At the same time, there have been changes in corporate culture, in the ways in which all companies do business. These are in addition to the changes in the chemical industry discussed in the previous section, and they are at least as important, because they alter the ways chemical engineers work. Two major changes are especially important: the way in which corporations organize their product design, and the ways in which corporate strategy affects jobs. Each is discussed below.

CORPORATE ORGANIZATION

The organization of product development is most easily discussed by a comparison of two limiting cases: organization by function, and organization by project. These are shown schematically in Figure 1.3–1. Both can be effective.

In a functional organization, different divisions have different responsibilities: marketing, research and development, engineering, legal affairs, and so on. Product development proceeds by each division doing its job, and then passing its results on to the next division. The result is like chemical reactions in series, as suggested by Figure 1.3–1. This organization is especially associated with large, established industrial companies that have major capital investments in manufacturing. For example, the marketing department of an automobile company could discover that consumers want better climate control, that is, better heating and air-conditioning. Marketing would report their results to research, who would develop the electronic controls required for this goal. Engineering would extend the research results so that the new controls could be manufactured cheaply and efficiently. Throughout the process of product design, the development is sequential: marketing talks largely to research, only rarely to engineering. Such a functional organization can be effective, but it is almost always slow.

A common alternative is a project organization. In a project organization, a core team is formed from the different divisions. The team will normally include representatives from marketing, research and development, engineering, production, and so on. These core team members will have complete responsibility – and a good deal of resources – to design and develop the target product. They will be judged not by their immediate functional supervisors, but rather by a panel of senior managers well versed in the company's long-term strategy. Functional supervisors still have the job of making the divisions run smoothly. Such divided management can be chaotic and inefficient. As Figure 1.3–1 suggests, it is like parallel chemical reactions, with a chance of synergy between functions. Above all, this form of product development is fast, and fast product development is believed to maximize profits, so that project management is currently the organization urged by most business consultants.

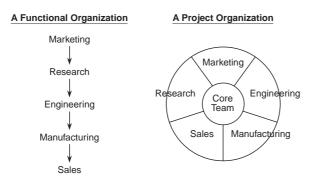


Figure 1.3–1. Two Limiting Types of Corporate Organization. The project organization is currently preferred because its speed and synergism outweigh its managerial complexity.

CORPORATE STRATEGY

Superimposed on its organization, a corporation will have strategic forces driving product development. Again, the driving forces are most easily described in terms of two limiting cases. First, corporations that look toward their markets for inspiration are said to be "market pull." Corporations that emphasize extending their service and technology are said to be "technology push." Examples of market-pull companies are common. W. M. Kellogg, the manufacturer of breakfast cereal, is interested in new products from grain. The company constantly assesses the market for consumer wishes for new cereals or new grain-based snack foods. Honeywell makes home thermostats, a major product because a significant fraction of the world's energy consumption goes for domestic heating. Honeywell is interested in any new products for home comfort that can complement their thermostats. Patagonia, the maker of technical mountain climbing equipment, now also makes raincoats. This organization is trying to expand its market: many more people need raincoats than ice axes.

Examples of technology-push companies are less common but can nonetheless be found among everyday names. W. L. Gore makes Goretex, that breathable film basic to high-quality raincoats. But Gore does not make raincoats: instead, this company has used its basic material to make medical products, including arterial transplants. Exxon-Mobil has used its knowledge of petrochemical reactions to develop a series of new metallocene catalysts for polyolefins. Astra-Zeneca has used its experience with injectable therapeutics to develop delivery systems for different drugs. Interestingly, both market-pull and technology-push companies can use the same product development procedure. This procedure is described next.

1.4 The Product Design Procedure

Product design is a major topic both in subjects such as sales and marketing, and in technical professions such as mechanical engineering. Not surprisingly, schemes for this design procedure vary widely. Many are complex, especially with respect to the role of management. Many have features that seem specific to the particular subdiscipline that they represent.

The product design procedure used in this book is a simplification and generalization of those used in these other areas. It depends on four steps:

- 1. *Needs*. What needs should the product fulfill?
- 2. Ideas. What different products could satisfy these needs?
- 3. Selection. Which ideas are the most promising?
- 4. *Manufacture*. How can we make the product in commercial quantities?

The characteristics of this approach are discussed in the rest of this section. We shall see as we go along that the application of this template to the case of chemical products leads to new features of the design process.

HOW THE PROCEDURE ORGANIZES THIS BOOK

These four steps are the key to the organization of this book. Assessment of needs, the subject of Chapter 2, includes deciding on a standard for comparison – a benchmark – and converting the qualitative needs to quantitative specifications. The benchmark chosen may be an existing product or an ideal. It must be as well defined as possible so that any specifications are definitive.

Finding ideas that might meet these needs is the next step in product design. Normally, we will wish to search for a large number of these ideas by all reasonable means. This search, the subject of Chapter 3, may include brainstorming by individuals and teams and synthesizing tangent compounds by combinatorial chemistry. Once these numerous ideas are identified, they must be screened by using objective and subjective judgments, also described in Chapter 3.

At this point, we should have reduced the large number of fragmentary ideas for products to a short list of promising candidates. Typically, this reduction will be about a factor of twenty: if we start with a hundred ideas, we should have about five survivors. We must now select the best one or two for further design and development. If the characteristics of each of the surviving ideas were directly comparable, this would be easy. They normally are not. For example, we might be sure that one idea will work well but be expensive; a competing idea might be cheap but we may be unsure if it will work. Deciding between these ideas includes risk management, as described in Chapter 4.

Finally, we must manufacture the product and estimate the costs involved. These efforts, described in Chapters 5, 6, and 7, are different from those expected for commodity chemicals, where we expect to use dedicated, optimized equipment that operates continuously. Here, we will normally use generic equipment, run in batch for a variety of specialty products.

This is different from traditional chemical engineering – and is exciting.

LIMITATIONS OF THE PROCEDURE

The four-step procedure outlined above is controversial. We should review the controversies now, so that we are prepared for exceptions and diversions in the practice of product design. The controversies cluster around three criticisms: that the procedure is not general, that management and not technology is the key, and that product design is part of chemical process design. Each controversy merits discussion; they are tackled below.

First, is the four-step procedure as outlined general? It is clearly a major simplification. Many business texts argue that such a procedure is universally applicable for any product in any industry. These texts are frequently written by business consultants eager to make money by applying their own standard template to specific problems. At the same time, many professional product developers argue that this or any procedure cannot represent the peculiarities of their own industry, that only those with particular interests can hope to be effective. Although there is clearly some truth in this argument, these product developers may be like those

who have denied that correlations of heat transfer could be used for food products because they were based on measurements for petrochemicals.

We believe that both sides of the debate have their merits. The four-step procedure used here is unquestionably an approximation. Certain techniques introduced in particular steps of the procedure can have value at other stages. For example, risk management, introduced in the selection step in Chapter 4, may have value in screening product ideas, explored in Chapter 3. It is unlikely that real product design will always be a simple sequential procedure as we suggest; iteration between stages is almost certain to be necessary. Still, we must begin somewhere, and the current procedure has been for us a sound and creative start. We suggest trying it; any necessary modifications quickly become obvious in specific cases. A framework in which the subject may be understood is an aid to learning in chemical product design just as an analagous template has been successfully applied for years in process design.

The second controversy is the claim that management, not technology, is key to product design. An irritating feature of most business books on product design is the extreme emphasis on the central role of management. The implication is that technology is always available if only the managers do their job properly (or at least do what the consultants say). These books on product design know no inconvenient constraints such as the second law of thermodynamics or the difference between mass and moles.

We believe that the application of technology is central to chemical product design. Product design governed only by management reminds us of a Sidney Harris cartoon showing a few managers and an engineer standing in front of a flip chart. Though the flip chart is covered with equations, pie charts, and organization charts, the engineer is pointing to one small box, which says:

"then a miracle occurs."

The engineer remarks:

"I'm having trouble with this part."

On reading books on the management of product design, we can feel all too much like that engineer in the cartoon. In this book, we want to make sure that technology is carefully considered.

The third controversy about this book is the assertion that the subject is already covered as part of the existing study of process design. This serious assertion is most easily tested by comparing our template for product design with an example of the intellectual hierarchy suggested for process design. One successful and powerful hierarchy, suggested by Douglas (1988), is summarized on the left side of Table 1.4–1. After deciding whether a process is batch or continuous, one then moves on to flow sheets of inputs and outputs which are almost always continuous. The initial flow sheets center on the stoichiometry. The next level of the hierarchy, which adds the recycles, often involves a discussion of the chemical reactions. Once these are established, one moves on to the separation trains and finally to the heat integration. All of this makes for a good course.

1.5 Conclusions

| Process Design | Product Design |
|-------------------------------------|---------------------------------|
| 1. Batch vs. Continuous Process | 1. Identify Customer Needs |
| 2. Inputs and Outputs | 2. Generate Ideas to Meet Needs |
| 3. Reactors and Recycles | 3. Select among Ideas |
| 4. Separations and Heat Integration | 4. Manufacture Product |

If we want to emphasize product design, we need to go beyond this hierarchy. We cannot simply substitute the search for a product for drug delivery into the process design hierarchy. Instead, the four-step hierarchy suggested earlier is shown on the right side of Table 1.4–1. After first identifying a corporate need, one generates ideas to fill this need. One then compares these alternatives and finally decides on a scheme for manufacture. The manufacturing includes all of the process design hierarchy.

Thus the important steps in product design anticipate those in process design. Product design implies a focus on the initial decisions around the choice of the product and implicitly de-emphasizes its manufacture. This shifts our efforts away from the common engineering calculations that have been our bread and butter for decades. The new emphasis includes subjects that have normally been left to those directly concerned with business. It is this combination of business and technology that is the subject of this book.

1.5 Conclusions

Product design is the procedure by which customer needs are translated into commercial products. This procedure, which precedes process design, is especially valuable for specialty chemicals. Such specialties are an important focus of the present-day chemical industry, which is evolving beyond commodities that have been the emphasis in recent decades.

In this book, the product design procedure is organized as four sequential steps. The first step, described in Chapter 2, is the identification of customer needs and the translation of the needs into product specifications. The second step, in Chapter 3, describes generating and winnowing ideas to fill these needs. In the third step, in Chapter 4, the best ideas are selected for commercial development. The last step, in Chapters 5–7, includes manufacture and economics. The result is a template for chemical product design.

We must stress that management, especially senior management, is much more likely to be involved in product design than in process design. As a member of a product team, each engineer or chemist will be involved in a management review at each stage of the design process. This review will be critical; that is, the review will decide on whether the project should continue. To reflect this, in the "conclusions" section of each chapter, we will mention this review, and discuss

what human interactions are likely. These human interactions are as important as the technology.

FURTHER READING

- Blessing, L. T. M. (1994). A *Process-Based Approach to Computer-Supported Engineering Design*. PhD Thesis, University of Twente, ISBN 0952350408.
- Cooper, R. G. (1993). Winning at New Products, Accelerating the Process from Idea to Launch, 2nd ed. Addison-Wesley, Reading, MA, ISBN 0201563819.
- Douglas, J. M. (1988). Conceptual Design of Chemical Processes. McGraw-Hill, New York, ISBN 0070177627.
- Graedel, T. E. and Allenby, B. R. (1996). *Design for Environment*. Prentice-Hall, Upper Saddle River, NJ, ISBN 0135316820.
- Kanter, R. M., Kao, J., and Wiersema, F. (eds.) (1997). *Innovation Breakthrough Thinking at 3M, DuPont, GE, Pfizer, and Rubbermaid*. Harper Collins, New York, ISBN 088730771X.
- Pahl, G. and Beitz, W. (1996). *Engineering Design, a Systematic Approach*, 2nd ed. Springer, New York, ISBN 3540199179.
- Spitz, P. (1988). Petrochemicals: The Rise of an Industry. Wiley, New York, ISBN 0471859850.
- Ulrich, K. T. and Eppinger, S. D. (2000). *Product Design and Development*, 2nd ed. McGraw-Hill, New York, ISBN 0071169938.