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978-1-107-01816-7 - Observational Molecular Astronomy: Exploring the Universe
Using Molecular Line Emissions

David A. Williams and Serena Viti

Excerpt

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Introduction

1.1 Why Are Molecules Important in Astronomy?

Molecules pervade the cooler, denser parts of the Universe. As a useful rule of thumb, cosmic gases at temperatures of less than a few thousand K and with number densities greater than one hydrogen atom per cm^3 are likely to contain some molecules; even the Sun's atmosphere is very slightly molecular in sunspots (where the temperature – at about 3200 K – is lower than the average surface temperature). However, if the gas kinetic temperatures are much lower, say about 100 K or less, and gas number densities much higher, say more than about 1000 hydrogen atoms per cm^3 , the gas will usually be almost entirely molecular. The Giant Molecular Clouds (GMCs) in the Milky Way and in other spiral galaxies are clear examples of regions that are almost entirely molecular. The denser, cooler components of cosmic gas, such as the GMCs in the Milky Way Galaxy, contain a significant fraction of the nonstellar baryonic matter in the Galaxy. Counterparts of the GMCs in the Milky Way are found in nearby spiral galaxies (see Figure 1.1). Although molecular regions are generally relatively small in volume compared to hot gas in structures such as galactic jets or extended regions of very hot X-ray-emitting gas in interstellar space, their much higher density offsets that disparity, and so compact dense objects may be more massive than large tenuous regions.

Such dense, cool regions are of course important in themselves, in adding to our description of the total content of galaxies. But they are also important for our understanding of how galaxies evolve because this denser, cooler gas is the only reservoir of matter for future star formation. Measuring the mass of this reservoir gas in a galaxy and comparing with the existing stellar mass may, for example, give some indication of the evolutionary state of that galaxy. Alternatively, the interaction of an outflow from an active galactic nucleus with cool dense gas in the galaxy can produce a signature chemistry that through

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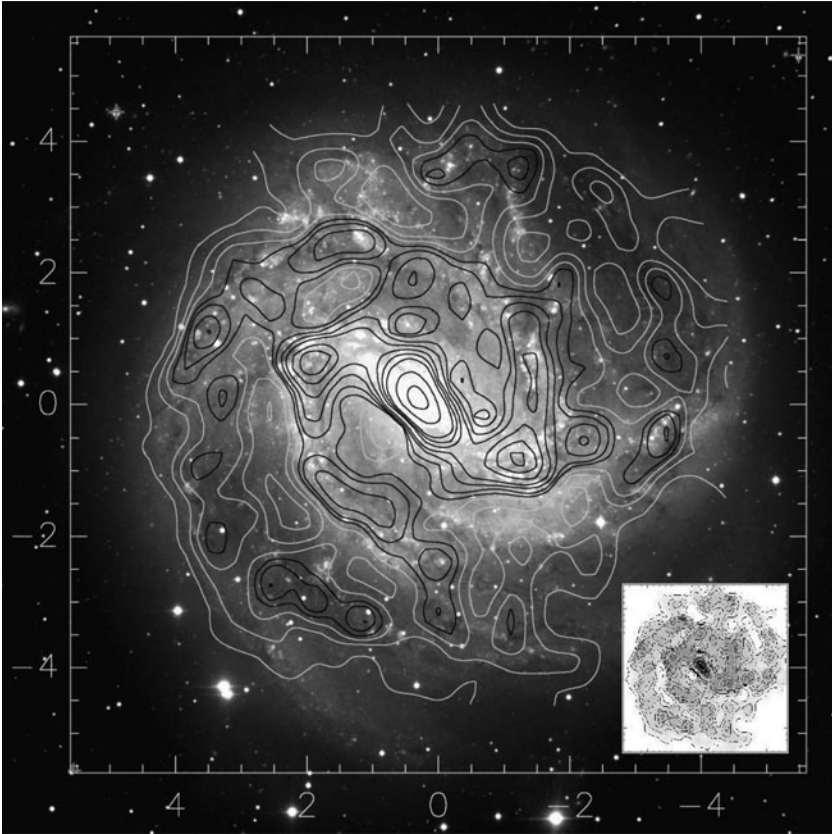
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Figure 1.1. Velocity-integrated CO ($J = 1-0$) intensity as contours superposed on a map centre of M83 produced from images in B, V, and R of M83, a nearby spiral galaxy with an estimated mass of $10^{10} M_{\odot}$. The x - and y -axes are offsets in RA and Dec from the centre of M83, measured in arcminutes. The average density in the GMCs is of the order of $100-400 \text{ cm}^{-3}$. The CO is associated with small regions of higher densities ($\leq 10^3-10^5 \text{ cm}^{-3}$) and temperatures of the order of 50 K, where star formation occurs. The inset shows the CO ($1-0$) map in grayscale. (Reproduced with permission from Lundgren, A. A., Wiklind, T., Olofsson, H., and Rydbeck, G. 2004. *Astronomy & Astrophysics* 413, 505.) Copyright ESO.

its specific molecular emissions may reveal important details of the outflow, such as its mass loss rate. No less important, but on a smaller spatial scale than GMCs, the collapse of gas from a tenuous state to a dense star-forming core can be followed by measuring line emissions from the molecular gas, even though the temperature may be as low as 10 K or even less. Indeed, the low temperature is maintained during much of the collapse by these molecular

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1.2 A Very Brief History of the Discovery of Molecules in Space 3

emissions and also by continuum emissions of the dust. At the end of that collapse process, the newly formed star irradiates any surrounding debris that was not incorporated into the star and generates a new chemistry that provides new molecular signatures. In particular, a protoplanetary disk surrounding the young star is the location in which planet formation occurs, and is also almost entirely molecular. The disk responds to the intense and growing radiation from the central star, and to its powerful wind, in processes that can generate new and useful diagnostic molecules.

Thus, many processes of topical interest in modern astronomy and astrophysics involve cold dense gas or the interaction of radiation or of violent processes with cold dense gas. This book is offered as a guide for astronomers who wish to use molecules as probes of these kinds of processes, and in particular to address the following main questions:

- What kinds of molecules arise in different astronomical situations?
- Which molecular species are the most useful tracers of gas in these different situations?
- Which molecular species are the most useful for determining important physical parameters (e.g., cosmic ray flux, local radiation field, elemental abundances, and so forth) in those situations?
- How does one convert basic observational data taken at the telescope to astrophysically useful information (e.g., column densities or fractional abundances) about an astronomical object?

1.2 A Very Brief History of the Discovery of Molecules in Space

Optical absorption lines, apparently molecular in origin, were first detected in 1937 in the spectra of bright stars, along lines of sight through the diffuse interstellar medium of the Milky Way. A few years later, on the basis of laboratory work, these and other lines were attributed to the diatomic radicals CH, CH⁺, and CN. No further detections were made until 1963, when OH masers were detected in the radio. Advances in detector technology permitted the development of millimetre wave and submillimetre wave astronomy and led to a veritable flood of new detections of molecular rotational transitions beginning in 1967. Some detections in other parts of the electromagnetic spectrum were also important. Molecular hydrogen, which has no dipole and therefore very weak rotational transitions, was first detected by a UV rocket experiment in 1970 by absorption in the Lyman and Werner bands; see Figure 1.2 for a

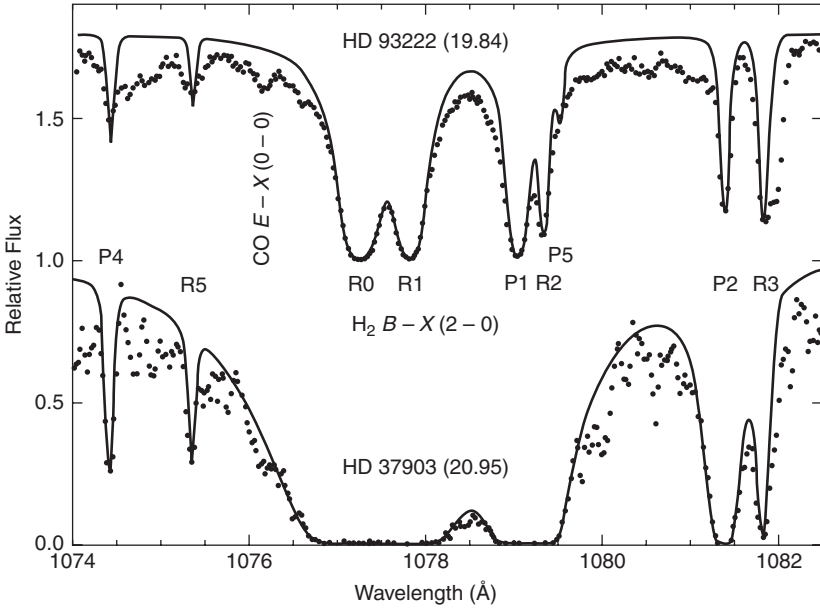


Figure 1.2. A piece of the UV absorption spectrum of H_2 towards two diffuse lines of sight taken with the space observatory Far Ultraviolet Spectroscopic Explorer (FUSE). These spectra show the electronic as well as ro-vibrational structure of the fundamental molecule H_2 . The solid lines represent model fits to the spectra. The main features are the $\text{B-X}(2-0)$ vibrational bands, and are labelled in the conventional notation (see Chapter 2). Here X represents the ground electronic state and B the first stable excited electronic state. The logarithm of the total hydrogen column density is indicated for each line of sight. (Reproduced by permission of the AAS from Sheffer, Y., Rogers, M., Federman, S. R., Abel, N. P., Gredel, R., Lambert, D. L., and Shaw, G. 2008. *Astrophysical Journal*, 687, 1075.)

recent detection. Molecular hydrogen is the seminal molecule for all interstellar chemistry, as we shall see.

From that time, the number of detected molecular species rose rapidly year by year and it soon became clear that the interstellar medium is a chemically complex environment (see e.g., Figure 1.3). An up-to-date list of detected molecular species is maintained at several websites (e.g., <http://www.astro.uni-koeln.de/cdms/molecules/>). A list of detected molecular species (as of 2012) organised by type of source is provided in Table 1.1.

Many isotopic varieties, in which, for example, D replaces H, or ^{13}C replaces ^{12}C , or ^{17}O or ^{18}O replaces ^{16}O , are also found, so that the total number of identified molecular species in interstellar and circumstellar space is very much larger than the total of main isotopes (which is currently ~ 180).

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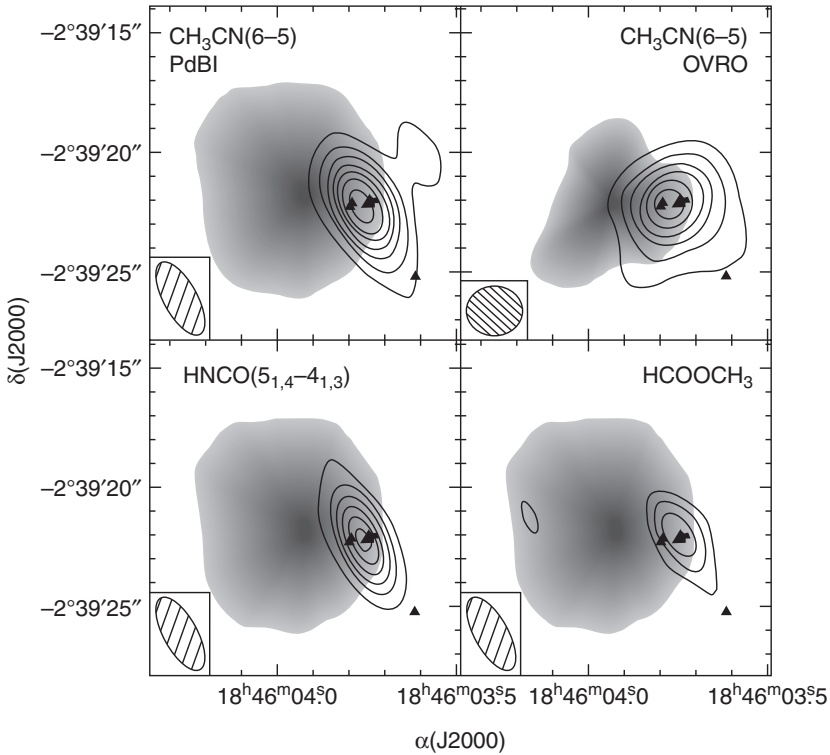


Figure 1.3. Spectra of complex molecules surrounding a massive star formation core, G29.96-0.02 (see also Chapter 5). The solid contours represent the molecular emissions and the grayscale indicates continuum emission from the ionised gas at 2.7 mm. (Reproduced, with permission, from Olmi, L., Cesaroni, R., Hofner, P., Kurtz, S., Churchwell, E., and Walmsley, C. M. 2003. *Astronomy & Astrophysics*, 407, 225.) Copyright ESO.

The first detections of extragalactic molecules were made in the 1970s. The current record for a molecular detection in a high-redshift galaxy is of CO at redshift $z = 6.42$ in 2003, in a gravitationally lensed quasar.

These detections attracted a great deal of attention, and the subjects of astrochemistry, bridging astronomy, chemistry, and physics emerged to try to account for the extraordinary range and variety of the detected species. However, there was no mere ‘stamp-collecting phase’ of molecular detections; in parallel with the development of astrochemistry, molecular emissions were immediately used to trace the existence of and physical conditions in interstellar and circumstellar gas. Such studies led to the discovery of previously unsuspected but important astronomical features – such as the GMCs in the inner part of the Milky Way. The structure of molecular outflows near cool stars

Table 1.1. List of detected molecular species with main regions in space where they have been observed

Molecule	Source	Molecule	Source	Molecule	Source
H ₂	dm, of	AlF	2 Atoms	AlCl	circ
CO	dm, circ, cc, yso, of, eg	C ₂	circ	CH	dm, eg
SH	dm	CH ⁺	dm	CN	dm, circ, eg
HCl ⁺	dm	CO ⁺	dm, eg	CP	circ
SiC	dm	HCl	circ, yso	KCl	circ
NH	dm, yso	NO	cc, yso, circ	NS	dm, yso, eg
NaCl	dm, eg	OH	cc, eg	PN	yso
SO	circ	SO ⁺	dm, circ, eg	SiN	circ
SiO	cc, yso, dm, circ, of, eg	SiS	dm, eg	CS	cc, yso, dm, circ, of, eg
HF	of, circ, yso, eg	O ₂	of, circ, yso	CF ⁺	dm
PO	dm, eg	AlO	dm, yso	OH ⁺	dm, eg
CN ⁻	circ	SH ⁺	circ		
	circ		dm		
C ₃	circ, dm	C ₂ H	3 Atoms	C ₂ O	cc
C ₂ S	cc, eg	CH ₂	yso, cc, dm, circ, eg	HCN	dm, cc, yso, circ, of, eg
HCO	cc, eg	HCO ⁺	yso, dm	HCS ⁺	cc, yso, dm
HOC ⁺	cc, eg	H ₂ O	cc, of, yso, eg	H ₂ S	cc, yso, of, circ, eg
HNC	eg	HNO	cc, yso, of, circ, eg	MgCN	circ
MgNC	dm, cc, yso, circ, of, eg	N ₂ H ⁺	yso	N ₂ O	dm
NaCN	circ	OCS	cc, yso, of	SO ₂	cc, yso, of, eg
SiC ₂	circ	CO ₂	cc, yso, of, eg	NH ₂	dm
H ₃ ⁺	circ	SiCN	yso	AlNC	circ
SiNC	dm	HCP	circ	CCP	circ
AlOH	circ	H ₂ O ⁺	circ	H ₂ Cl ⁺	dm, yso
KCN	circ	FeCN	dm, yso, eg		
	circ		circ		

H ₂ O ₂	dm	C ₃ H	4 Atoms	C ₃ N	dm
C ₃ O	dm	C ₃ S	cc, circ, eg	C ₂ H ₂	cc, circ
NH ₃	dm, cc, yso, of, eg	HCCN	ccc, circ	HCNH ⁺	cc
HNCO	yso, eg	HNCS	circ	HOCO ⁺	dm, yso
H ₂ CO	dm, cc, yso, of, eg	H ₂ CN	cc	H ₂ CS	cc, yso, of, circ, eg
H ₃ O ⁺	yso, eg	SiC ₃	circ	CH ₃	dm
C ₃ N ⁻	circ	HCNO	yso, eg	HOCN	dm, yso
HSCN	yso				
C ₅	circ		5 Atoms		
C ₃ H ₂	cc, yso, circ, eg	C ₄ H	dm, cc, circ	C ₄ Si	circ
HC ₃ N	cc, circ, eg	H ₂ CCN	cc, yso, circ	CH ₄	circ
H ₂ CNH	dm, yso	HC(O)CN	yso	HCOOH	cc, yso
C ₄ H ⁻	circ	H ₂ C ₂ O	dm, cc, yso	H ₂ NCN	yso
		SiH ₄	circ	H ₂ COH ⁺	dm, cc
C ₂ H ₄	circ		6 Atoms		
CH ₃ NC	yso	CH ₃ CN	cc, yso, of, eg	HC ₄ H	circ, eg
C ₃ H	circ, cc	H ₂ C ₄	circ, cc, yso	CH ₂ CNH	yso
C ₅ N	circ, cc	HC ₃ NH ⁺	cc	CH ₃ C ₂ H	cc, yso
HC ₂ CHO	yso	HC ₄ N	circ	CH ₃ OH	cc, yso, eg
NH ₂ CHO	yso	C ₃ H ₂ O		CH ₃ SH	yso
C ₆ H	circ, cc, yso		7 Atoms		
C ₂ H ₃ CN	cc, yso, eg	C ₆ H	circ, cc, yso	CH ₃ NH ₂	yso
C ₂ H ₃ OH	yso	HC ₅ N	circ, cc	CH ₃ CHO	cc, yso, eg
		CH ₂ OCH ₂	yso		

(cont.)

Table 1.1 (*cont.*)

Molecule	Source	Molecule	Source	Molecule	Source
H ₂ C ₆	circ, cc, yso	HC ₆ H	8 Atoms circ, eg	C ₇ H	circ, cc
CH ₃ C ₃ N	cc	CH ₂ CCHCN	cc	NH ₂ CH ₂ CN	yso
HCOOCH ₃	yso, of	CH ₃ COOH	yso, of	HOCH ₂ CHO	yso
C ₂ H ₃ CHO	yso				
CH ₃ C ₄ H	cc	CH ₃ CHCH ₂	9 Atoms cc	C ₈ H	circ, cc
HC ₇ N	circ, cc	C ₈ H	circ, cc	C ₂ H ₅ CN	yso
CH ₃ CONH ₂	yso	C ₂ H ₅ OH	yso, of	CH ₃ OCH ₃	yso
CH ₃ C ₅ N	cc	CH ₃ COCH ₃	10 Atoms yso	HOCH ₂ CH ₂ OH	yso
C ₂ H ₅ CHO	yso				
CH ₃ C ₆ H	cc	HC ₉ N	11 Atoms circ, cc	HCOOC ₂ H ₅	yso
C ₆ H ₆	circ, eg	C ₃ H ₇ CN	12 Atoms yso		
HC ₁₁ N	circ, cc		13 Atoms		

Abbreviations: dm = diffuse medium (including translucent clouds); circ = circumstellar envelope around evolved star/protoplanetary nebula; cc = cold cloud core; yso = gas around a young stellar object, including observations of the hot core in the galactic centre; of = outflow; eg = extragalactic regions. Some of the abbreviations used in this list are taken from E. Herbst and E. F. van Dishoeck. 2009. *Annual Review of Astronomy and Astrophysics*, 47: 427. We do not include isotopologues in this table.

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Table 1.2. Types of interstellar and circumstellar region and their physical characteristics

Region	n_{H} (cm^{-3})	T (K)
Coronal gas	$<10^{-2}$	5×10^5
HII regions	>100	1×10^4
Diffuse gas	100–300	70
Molecular clouds	10^4	10
Prestellar cores	10^5 – 10^6	10–30
Star-forming regions	10^7 – 10^8	100–300
Protoplanetary disks	10^4 (outer)– 10^{10} (inner)	10(outer)–500(inner)
Envelopes of evolved stars	10^{10}	2000–3500

All of the regions, except coronal gas and HII regions, can be probed with molecules.

was revealed, and molecular ices were found to be present in the interstellar medium.

Molecular emissions, along with X-ray, UV, optical, and infrared emissions, have helped to define the variety of physical states of interstellar gas. These range over at least a factor of $\sim 10^{12}$ in density and $\sim 10^5$ in temperature, from number densities of $\sim 10^{-2} \text{ cm}^{-3}$ and temperatures $\sim 10^6$ K in so-called coronal gas to values of $\sim 10^{10} \text{ cm}^{-3}$ and ~ 10 K in protoplanetary disks. Table 1.2 lists the known interstellar and circumstellar components. Of these types of region, diffuse clouds, molecular clouds, prestellar cores, star-forming regions, protoplanetary disks, circumstellar envelopes, and the ejecta of novae and supernovae can be studied through molecular emissions.

As astronomy moves into a new phase dominated by data from revolutionary space- and ground-based instrumentation, molecular astronomy is no longer a semidetached specialty of work in the millimetre and submillimetre regions of the spectrum. Molecular astronomy now addresses questions at the forefront of the subject, and is simply part of the range of expertise that astronomers must command. This book is intended to help astronomers become equally skilled in molecular line observations as in making observations in other regions of the spectrum.

1.3 Gas and Dust

1.3.1 Gas Composition for Interstellar Chemistry

The raw material for our considerations of chemistry consists of gas and dust. The gas consists mainly of hydrogen and helium with a small component

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Table 1.3. Approximate solar elemental abundances relative to the total number of hydrogen nuclei

Element	Abundance
H	1
He	9×10^{-2}
O	5×10^{-4}
C	3×10^{-4}
N	7×10^{-5}
Si	3×10^{-5}
Mg	4×10^{-5}
Fe	3×10^{-5}
S	1×10^{-5}
Na, Ca	2×10^{-6}

Note that solar elemental abundances may not be valid for all regions of space.

of other elements formed in stellar nucleosynthesis and distributed by novae and supernovae and by stellar winds. Obviously, the ability of a gas to form molecules involving carbon, oxygen, nitrogen, sulfur, and other elements (as well as hydrogen) depends on the abundance of the small component of other elements relative to hydrogen. These relative elemental abundances may vary from place to place within a galaxy and from galaxy to galaxy. Solar abundances are often used as a conventional reference level; solar elemental abundances relative to hydrogen are shown in Table 1.3. Gas with these relative elemental abundances is said to have solar *metallicity*. The metallicity is an important parameter in astrochemistry; we consider the effect on the chemistry of varying the metallicity in Chapter 4. It is often assumed that although the metallicity may vary, the abundances of the elements relative to each other follow solar values. However, this may not be the case everywhere. For example, if considering the early Universe, supernovae of different masses may lead to quite different predictions of relative abundances of the major elements carbon, nitrogen, and oxygen. Stellar evolution models for initially zero-metallicity gas predict nitrogen to be underabundant whereas oxygen and magnesium are overabundant compared to solar metallicity. Also, dredge-up processes in evolved stars are observed to create distinct differences in elemental abundances. For example, some stars may have different C:O ratios in their atmospheres and envelopes