Groundwater in Geologic Processes

Second Edition

Interest in the role of *Groundwater in Geologic Processes* has increased steadily over the past few decades. Hydrogeologists and geologists are now actively exploring the role of groundwater and other subsurface fluids in such fundamental geologic processes as crustal heat transfer, ore deposition, hydrocarbon migration, earthquakes, tectonic deformation, diagenesis, and metamorphism.

Groundwater in Geologic Processes is the first comprehensive treatment of this body of inquiry. Chapters 1 to 4 develop the basic theories of groundwater motion, hydromechanics, solute transport, and heat transport. Chapter 5 applies these theories to regional groundwater flow systems in a generic sense, and Chapters 6 to 13 focus on particular geologic processes and environments. Relative to the first edition of *Groundwater in Geologic Processes*, this second edition includes a much more comprehensive treatment of hydromechanics (the coupling of groundwater flow and deformation). It also includes new chapters on "compaction and diagenesis," "metamorphism," and "subsea hydrogeology." Finally, it takes advantage of the substantial body of published research that has appeared since the first edition in 1998.

The systematic presentation of theory and application, and the problem sets that conclude each chapter, make this book ideal for undergraduate- and graduate-level geology courses (assuming that the students have some background in calculus and introductory chemistry). It also serves as an invaluable reference for researchers and other professionals in the field.

STEVE INGEBRITSEN has been a member of the US Geological Survey since 1980 and received M.S. (1983) and Ph.D. (1986) degrees in hydrogeology from Stanford University. He currently serves as a research Branch Chief in the USGS Water Resources Discipline. Steve has received the John Hem Excellence in Science and Engineering Award from the National Groundwater Association (2000), served as the Geological Society of America's Birdsall-Dreiss Distinguished Lecturer (2001), and received the O. E. Meinzer Award from GSA (2003).

WARD SANFORD has been a hydrogeologist with the US Geological Survey since 1983. He received a B.S. (1983) with honors and highest distinction from Purdue University and a Ph.D. (1987) from Penn State University. Ward is currently research advisor for the groundwater hydrology

discipline of the USGS National Research Program. He has received the Geological Society of America Young Scientist Award (Donath Medal) (1995) and the John Hem Excellence in Science and Engineering Award from the National Groundwater Association (2000).

CHRIS NEUZIL has been a hydrogeologist with the US Geological Survey since 1977. He holds a B.A. in geology from the State University of New York at Binghamton (1971) and a Ph.D. in hydrogeology from Johns Hopkins University (1980). Chris is a senior scientist in the USGS National Research Program. He received the O. E. Meinzer Award from the Geological Society of America (1991) and served as GSA's 1995 Birdsall-Dreiss Distinguished Lecturer. He serves or has served on the editorial boards of *Water Resources Research, Ground Water*, and *Geofluids*.

Groundwater in Geologic Processes Second Edition

Steven E. Ingebritsen

United States Geological Survey Menlo Park, California

Ward E. Sanford

United States Geological Survey Reston, Virginia

Christopher E. Neuzil

United States Geological Survey Reston, Virginia



> CAMBRIDGE UNIVERSITY PRESS Cambridge, New York, Melbourne, Madrid, Cape Town, Singapore, São Paulo

Cambridge University Press The Edinburgh Building, Cambridge CB2 2RU, UK

Published in the United States of America by Cambridge University Press, New York

www.cambridge.org Information on this title: www.cambridge.org/9780521603218

© Cambridge University Press 1998 © S. Ingebritsen, W. Sanford and C. Neuzil 2006

This publication consists of United States Government contributions for which no copyright coverage shall exist in the United States.

This publication is in copyright. Subject to statutory exception and to the provisions of relevant collective licensing agreements, no reproduction of any part may take place without the written permission of Cambridge University Press.

First published 1998 First paperback edition 1999 Second edition 2006

Printed in the United Kingdom at the University Press, Cambridge

A catalog record for this publication is available from the British Library

Library of Congress Cataloging in Publication data

ISBN-13 978-0-521-60321-8 paperback ISBN-10 0-521-60321-8 paperback

Cambridge University Press has no responsibility for the persistence or accuracy of URLs for external or third-party internet websites referred to in this publication, and does not guarantee that any content on such websites is, or will remain, accurate or appropriate.

Contents

	Pref	page xiii		
	Ack	nowledg	gements	xviii
	List	XX		
1	Gro	undwa	ter flow	1
	1.1	Darcy	's law	2
		1.1.1	The limits of Darcy's law	4
		1.1.2	Driving forces for groundwater flow	6
	1.2	Crusta	al permeability	10
		1.2.1	Permeability versus porosity	10
		1.2.2	Heterogeneity and anisotropy	12
		1.2.3	Scale dependence	15
		1.2.4	Depth dependence	18
		1.2.5	Time dependence	21
		1.2.6	Some limiting values	22
	1.3	Conce	eptualizing groundwater systems	24
	1.4	The co	ontinuum approach	26
	1.5	The g	roundwater flow equation	28
		1.5.1	Conservation of mass	28
		1.5.2	The storage term	31
		1.5.3	Various forms of the groundwater flow	
			equation	33
		Proble	ems	36
2	Hyd	Iromec	hanical coupling	38
	2.1	Hydro	omechanical equations	39
		2.1.1	Elastic strain	43
		2.1.2	Effective stress	45
		2.1.3	Constitutive law for poroelasticity	50
		2.1.4	Force equilibrium and the equations of	
			poroelastic deformation	52
		2.1.5	The groundwater flow equation for	
			poroelastic media	56

vi Contents

		2.1.6	1 8	
			poroelastic media	61
	2.2	Inelas	tic deformation and finite strain	66
		2.2.1		67
		2.2.2		
			deformation	75
	2.3	Uncer	tainty in hydromechanical descriptions	76
		2.3.1		76
		2.3.2	Uncertain boundary and initial conditions	79
	2.4	Pragn	natism in analysis: useful simplifications	80
			Purely vertical strain	80
		2.4.2	Numerical tricks and extensions of	
			poroelasticity	83
		2.4.3	Porosity and stress gradients as indicators of	
			long-term strain	84
	2.5	Under	rstanding hydromechanical phenomena	86
		Proble	ems	87
3	Solu	ite tran	isport	89
	3.1	Gover	rning equations	89
		3.1.1	Molecular diffusion	90
		3.1.2	Advection	92
		3.1.3	Mechanical dispersion	93
		3.1.4	Mass-balance equation	96
		3.1.5	Chemical reactions	97
		3.1.6	Initial and boundary conditions	99
	3.2	Nume	rical solution techniques	101
	3.3	Densi	ty-driven flow	104
	3.4	Multi	component diffusion	105
	3.5	Multi	component reactive transport	108
		3.5.1	Rate-based reactions	111
		3.5.2	Surface reactions	113
		3.5.3	Homogeneous reactions	114
		3.5.4	Heterogeneous reactions	116
		3.5.5	Solution algorithms	117
	3.6		iltration	119
	3.7	The ro	ole of microbes	120
		Proble	ems	122
4	Hea	t trans	port	124
	4.1		rning equations	124
		4.1.1	Pressure and enthalpy as dependent	
			variables	125

Cambridge University Press
0521603218 - Groundwater in Geologic Processes, Second Edition
Steven E. Ingebritsen, Ward E. Sanford and Christopher E. Neuzil
Frontmatter
Moreinformation

			Contents	vii
	4.1.2		129	
	4.1.3			
		compressible fluids	131	
	4.1.4		133	
	4.1.5	6 6	124	
	410	equations	134	
	4.1.6	Extending the governing equations to three dimensions	125	
	417		135	
		Assumptions	136	
		Fluid properties Numerical solution	138	
4.2	4.1.9		139	
4.2		and boundary conditions	139 141	
4.3	-	erature-based formulations	141	
4.4		imensional groundwater flow	143	
		Steady vertical flow	145	
4.5		Flow in a confined aquifer or fault zone	140	
4.5		Nusselt number	148	
		Peclet number	148	
		Rayleigh number	149	
4.6		ncy-driven flow	150	
4.7	Heat p	•	151	
4.8	-	rsion of gravitational potential energy to heat	152	
ч.0	Proble		155	
			155	
-		cale flow and transport	157	
5.1	-	nal groundwater flow	158	
5.2		alous fluid pressures	159	
		Geologic forcing	161	
		Elevated fluid pressures	163	
	5.2.3		164	
	5.2.4		165	
5.3	-	nal-scale solute transport	168	
	5.3.1	Groundwater age	168	
	5.3.2	0 1	172	
	5.3.3	Evolution of regional groundwater		
		chemistry	174	
5.4	-	nal-scale heat transfer	177	
	5.4.1	The conductive regime in sedimentary		
	.	basins	177	
	5.4.2	Thermal effects of groundwater flow in		
		sedimentary basins	179	
	5.4.3	Some case studies of sedimentary basins	181	

5

viii Contents

		5.4.4	An example from volcanic terrane	184
		5.4.5	The stress – heat flow paradox of the San	
			Andreas fault	187
		Proble	ems	191
6	Ore	deposi	ts	193
	6.1	-	ssippi Valley type deposits	194
		6.1.1	Evidence for regional-scale brine migration	195
		6.1.2	The role of brines	197
		6.1.3	Controls on ore deposition	198
		6.1.4	Driving forces for fluid flow	199
		6.1.5	Unresolved issues with United States MVTs	202
		6.1.6	The Irish "MVTs"	203
	6.2	Other	stratiform base-metal deposits	205
	6.3	Sedim	nent-hosted uranium	207
		6.3.1	Redox control of uranium solubility	207
		6.3.2	Tabular uranium deposits	208
		6.3.3	Unconformity-type uranium deposits	208
	6.4	Miner	alization through in situ diagenesis	213
		6.4.1	Supergene enrichment of porphyry copper	213
		6.4.2	Colombian emeralds	216
		Proble	ems	218
7	Hyd	lrocarb	ons	220
	7.1	Thern	nal maturation	221
		7.1.1	The oil window	221
		7.1.2	Groundwater flow and the thermal regime	222
	7.2	Migra	tion	226
		7.2.1	Capillary effects	226
		7.2.2	Primary migration	227
		7.2.2 7.2.3		227 228
	7.3	7.2.3	Secondary migration	
	7.3 7.4	7.2.3 Entrap	Secondary migration	228
		7.2.3 Entrap Gover flow	Secondary migration oment rning equations for immiscible multiphase	228 230 234
		7.2.3 Entrap Gover flow Case s	Secondary migration oment rning equations for immiscible multiphase studies	228 230 234 235
	7.4	7.2.3 Entrap Gover flow Case s	Secondary migration oment rning equations for immiscible multiphase	228 230 234 235 236
	7.4 7.5	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin	228 230 234 235 236 239
	7.47.57.6	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2 Pressu	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin ure regimes in hydrocarbon basins	228 230 234 235 236 239 241
	7.4 7.5	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2 Pressu Hydro	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin ure regimes in hydrocarbon basins ocarbon resources	228 230 234 235 236 239 241 245
	7.47.57.6	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2 Pressu	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin ure regimes in hydrocarbon basins ocarbon resources	228 230 234 235 236 239 241
8	7.47.57.67.7	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2 Pressu Hydro Proble	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin ure regimes in hydrocarbon basins ocarbon resources	228 230 234 235 236 239 241 245
8	7.47.57.67.7	7.2.3 Entrap Gover flow Case s 7.5.1 7.5.2 Pressu Hydro Proble	Secondary migration oment rning equations for immiscible multiphase studies The Uinta basin The Los Angeles basin ure regimes in hydrocarbon basins ocarbon resources ems	228 230 234 235 236 239 241 245 246

Contents	ix
0011101110	

	8.1.2	Lateral and vertical variations	250
	8.1.3	Perturbations due to groundwater flow	252
8.2	Magn	natic-hydrothermal systems	253
	8.2.1	Magmatic heat sources	254
	8.2.2	Heat transfer from magma to groundwater	256
	8.2.3	Fluid circulation near magma bodies	257
	8.2.4	Permeabilities in near-magma environments	260
	8.2.5	Magnitude and time-variation of	
		hydrothermal discharge	262
8.3	Fluid	flow and heat transport near the critical point	267
	8.3.1	One-dimensional pressure-enthalpy paths	268
	8.3.2	Two-dimensional convection	270
8.4	Multi	phase processes	271
	8.4.1	Phase separation	272
	8.4.2	Vapor-dominated zones	273
	8.4.3	Pressure transmission	276
	8.4.4	Boiling point – depth curves	277
8.5	Hot sp	orings	279
8.6	Geyse		281
8.7		ermal resources	284
8.8		othermal ore deposits	285
	Proble	ems	287
Ear	thquak	24	289
	-	ive stress	290
		mb's law of failure	290
9.3		ed seismicity	295
2.0	9.3.1	The Rocky Mountain arsenal	296
		Rangely, Colorado	298
	9.3.3		300
9.4		pressures and tectonism	300
		Hubbert and Rubey	302
		Irwin and Barnes model for the San Andreas	304
	9.4.3	Byerlee and Rice models for the San	
		Andreas	306
	9.4.4	Earthquake swarms driven by deep, natural	
		fluid sources	309
9.5	Seism	icity modulated by shallow hydrology	311
9.6		quake-induced hydrologic phenomena	312
		6.1 Well behavior	313
	9.	6.2 Geysering	316
	9.	6.3 Streamflow changes	318

9

x Contents

	9.7	Stress,	earthquakes, and crustal permeability	322
		Probler	ns	324
10	Fyan	orites		325
10	10.1		ite formation	325
	10.1	-	The marine evaporite problem	325
		10.1.1	* *	328
			CaCl ₂ brines	330
			Magnesium depletion	332
			Continental evaporites	333
			Sabkhas	334
			Groundwater outflow	336
	10.2		ite burial and dissolution	340
	10.3	*	c rise of evaporites	342
		10.3.1	Variable-density convection	343
			Caprock formation	346
		Probler	1	349
	C			2.52
11		-	and diagenesis	352
	11.1	Compa		353
		11.1.1	Vertical loading and elevated fluid	257
		11 1 2	pressures Delege de composition	357
			Delayed compaction	360
			Erosional unloading	364
			Tectonic compression	365
	11.2		Variations in fluid pressure	368
	11.2	0		371
		11.2.1	ı ı ı	371
		11.2.2	8	373
			Mechanochemical processes	378
		Probler	Mineral banding	381 383
		Problei	lis	202
12	Meta	morphis	sm	385
	12.1		e of fluids in metamorphism	386
		12.1.1	Evidence for voluminous fluid fluxes	386
		12.1.2	Sources of fluid	387
		12.1.3	Fluid dynamics	390
	12.2	Crustal	-scale permeability estimates	392
		12.2.1	Constraints from first-order calculations	393
		12.2.2	A permeability-depth curve based on	
			metamorphic data	394
		12.2.3	Implications for the brittle-ductile	
			transition	398

CAMBRIDGE

Cambridge University Press 0521603218 - Groundwater in Geologic Processes, Second Edition Steven E. Ingebritsen, Ward E. Sanford and Christopher E. Neuzil Frontmatter More information

Contents	xi

		12.2.4	Implications for fault behavior	399
	12.3	Heat ar	nd solute transport during metamorphism	401
	12.4	A mode	el of crustal-scale hydrology	403
	12.5	Metam	orphic environments	404
		12.5.1	Contact metamorphism	405
		12.5.2	Regional-contact metamorphism	
			(magmatic arcs)	407
		12.5.3	Regional metamorphism	
			(continent-continent collision)	409
		12.5.4	High-pressure metamorphic belts	
			(subduction zones)	409
		Problem	ns	410
13	Subs	ea hydro	ogeology	412
	13.1	•	versus subaerial hydrogeology	413
	13.2		permeability structure	415
	13.3		-driven subsea flow	418
	13.4	Hydrot	hermal circulation near the mid-ocean	
		ridge		420
		13.4.1	Importance to the Earth's thermal	
			budget	423
		13.4.2	Influence on ocean chemistry	424
		13.4.3	Properties of H ₂ O-NaCl fluids	425
		13.4.4	Quantitative description of MOR	
			systems	428
	13.5	Off-axi	s circulation and the role of seamounts	431
	13.6	Gas hy	drates	432
	13.7	Accreti	onary prisms and subduction zones	434
	13.8	Nearsh	ore hydrogeology	440
		13.8.1	Dolomitization of carbonate platforms	441
		13.8.2	Mixing-zone dissolution in carbonate	
			platforms	445
		13.8.3	Offshore groundwater flow and heat	
			transport	449
	13.9	Subduc	tion, metamorphism, and the world ocean	451
		Probler	ns	454
	Refer	ences		456
	Index			527
				/

Preface

Historically, interest in groundwater and other subsurface fluids was confined to a few specific disciplines in the Earth sciences, notably groundwater hydrology, soil physics, engineering geology, petroleum geology, and petroleum engineering. These disciplines tended to be "applied" in nature, with practitioners concentrating on the immediate and practical problems of water supply, water quality, mine dewatering, deformation under structural loads, and the location and recovery of fluid hydrocarbons.

This situation has changed over the past few decades. Hydrogeologists and geologists are now actively exploring the role of groundwater and other subsurface fluids in such fundamental geologic processes as crustal heat transfer, ore deposition, hydrocarbon migration, earthquakes, tectonic deformation, diagenesis, and metamorphism. *Groundwater in Geologic Processes* is the first comprehensive treatment of this body of inquiry. Relative to the first edition of *Groundwater in Geologic Processes*, this second edition includes a much more comprehensive treatment of hydromechanics (the coupling of groundwater flow and deformation). It includes new chapters on "compaction and diagenesis," "metamorphism," and "subsea hydrogeology." It also takes advantage of the substantial body of published research that has appeared since the first edition in 1998.

The growth of interest in the role of groundwater in geologic processes is indicated by the popularity of regularly scheduled meetings on "Water–Rock Interaction" and "Geofluids", by Penrose Conferences such as "Continental Magmatic-Hydrothermal Systems" (2004), and by an increasing number of related symposia at annual meetings of the American Geophysical Union and the Geological Society of America. Other indices of interest include the success and longevity of an annual, informal "Hubbert Quorum" on groundwater and geologic processes and the birth of a new journal dedicated to *Geofluids*.

We have attempted an integrated and systematic accounting of this body of inquiry, suitable for use as an adjunct to more-or-less general curricula in hydrogeology and geology, or as the primary text for courses on groundwater and geologic processes. The books that come closest to

CAMBRIDGE

Cambridge University Press 0521603218 - Groundwater in Geologic Processes, Second Edition Steven E. Ingebritsen, Ward E. Sanford and Christopher E. Neuzil Frontmatter <u>More information</u>

xiv Preface

addressing the intended audience (*Fluids in the Earth's Crust* by Fyfe and others, 1978; *Geology and Water* by Chapman, 1981; *Physical and Chemical Hydrogeology* by Domenico and Schwartz, 1990; *Introduction to Hydrogeology* by Deming, 2002; and *Geodynamics* by Turcotte and Schubert, 2002) in fact touch on only a few of the couplings between geology and groundwater, or are specific to particular hydrogeologic environments (e.g., *Hydrogeology of the Ocean Lithosphere* by Davis and Elderfield, 2004).

This book describes how the occurrence and movement of groundwater affects a wide range of geologic processes in diverse environments. We have defined "groundwater" in a broad sense, as any subsurface, aqueous fluid, including those of meteoric, connate, and magmatic origin. We originally considered naming this book *Fluids and Geologic Processes*. The term "fluids" is more inclusive but also tends to imply full consideration of an important nonaqueous crustal fluid, magma. Here, although we consider magma as a source of heat, fluids, and solutes to groundwater systems, we do not treat the dynamics of the magma itself.

The organization of this book is intended to facilitate use by both students and professionals. Chapters 1 to 4 develop the basic theories of groundwater motion, hydromechanics, solute transport, and heat transport. These four chapters comprise the theoretical core of the book, and we expect that most course instructors would elect to assign them during the first few weeks of a quarter or semester. Chapter 5 applies these theories to regional groundwater flow systems in a generic sense, and Chapters 6 to 13 focus on particular geologic processes and environments. Practicing earth scientists who are interested in particular geologic problems might turn to the latter part of the book first and then refer back to Chapters 1 to 4 as necessary to review pertinent theory.

We expect that when this book is used as a textbook, it will most often be in the context of a graduate-level course, and that many of the students will already have taken an introductory groundwater course. However, it has also proven accessible to undergraduate earth science students who have some understanding of calculus and introductory chemistry and physics but lack previous formal training in hydrogeology.

"Everything is coupled"

In order to treat the role of groundwater in geologic processes, we must first develop the pertinent theory, including equations of groundwater motion (Chapter 1) and descriptions of the couplings with deformation, solute transport, and heat transport. The coupling of groundwater flow with solute transport (Chapter 3) is usually included in standard developments of groundwater theory, because of its importance to water-quality

Preface xv

and contamination issues. However, the coupling with heat transport (Chapter 4) is often neglected. In most hydrogeologic practice it is fairly standard (and reasonable) to assume that groundwater flow takes place in an isothermal, nondeforming medium. For the geologic applications considered here, these assumptions often are not tenable.

Over the large spatial scales that pertain in many geologic applications, the assumption of isothermal flow (and thus constant fluid properties) is often poor. Given a typical geothermal gradient of 25 °C/km, temperatures in many large sedimentary basins may be expected to vary by 100 °C or more. For pure water, a 100 °C variation in temperature can cause a 4% variation in fluid density (greater than the density difference between freshwater and seawater) and a 600% variation in fluid viscosity (equivalent to a sixfold variation in hydraulic conductivity). In moderate- to high-temperature hydrothermal systems, much larger variations in fluid properties occur. For example, there is a threefold variation in the density of pure liquid water between room temperature and its critical point (374 °C, 22.06 MPa), and in multiphase (liquid–steam) systems, density variations range up to several orders of magnitude.

Furthermore, over the large time scales that pertain in many geologic problems, deformation often will be substantial. In the first edition of Groundwater in Geologic Processes our theoretical treatment of coupled groundwater flow and transport was fairly complete, but description of the coupling between groundwater flow and deformation was rudimentary. We generally neglected deformation except in deriving specific storage (Section 1.5.2) and in discussing sediment dewatering (Chapter 5.2), the "stress-heat flow paradox" of the San Andreas fault (Section 5.4.5), and earthquakes (Chapter 9). This is in accordance with standard hydrogeologic practice, but for some geologic problems a more general and rigorous treatment of coupled flow and deformation based on poroelastic theory is needed. In the first edition we did not elucidate this theory but instead referred the interested reader to other excellent references (e.g., Biot, 1941; Rice and Cleary, 1976; Roeloffs, 1996; Wang, 2000). For this second edition, in order to provide a more complete treatment of coupled groundwater flow and deformation, we enlisted the participation of Chris Neuzil (e.g., Neuzil, 2003).

In some environments the deformation rate will actually exceed the groundwater flow rate. Thus, for instance, in the rapidly deforming *accretionary prisms* of sediment that overlie many subducting oceanic plates, the net direction of fluid flow relative to fixed coordinates may be very different from the flow direction relative to the sedimentary matrix (Section 13.7). Deformation can also reduce porosity, thereby increasing fluid pressure, decreasing the effective stress, and reducing the amount of work required for tectonism. Deformation-related forcing xvi Preface

of groundwater flow is particularly effective in low-permeability strata (Section 5.2).

From the theoretical point of view, then, a unique aspect of the study of groundwater and geologic processes is the importance of the couplings of groundwater flow with heat transport and deformation, in addition to solute transport. Another unusual consideration (relative to standard hydrogeologic practice) is that every one of these couplings may be important to a given problem. Hydrothermal circulation near mid-oceanridge (MOR) hydrothermal vents provides an important and clear example (Section 13.4). We can assume that the fluid flow is governed by some form of Darcy's law, except perhaps very near some of the vents, where flow rates may be high enough that some energy is lost to turbulence. We would need to invoke a relatively complex, multiphase form of Darcy's law (Section 4.1.3), because large gradients in salinity between vents indicate that active phase separation (Section 8.4.1) occurs between a relatively dense, saline brine and a less saline vapor. Large gradients in salinity and temperature (approximately 2 °C to 400 °C) dictate that any complete model of the system must include both solute and heat transport. Furthermore, we must expect that the flow systems are highly transient; the exceptionally high rates of heat discharge from individual vents can only be explained as the result of rapid crystallization and cooling of large volumes of magma (Section 8.2), so that the intensity and spatial distribution of heat sources must vary with time. We would also expect precipitation and dissolution of minerals to cause continuous variations in porosity and permeability, because the extreme variations in fluid composition and temperature make for a highly reactive chemical environment. As a result of these transient phenomena, deformation enters the picture: as permeability, flow rates, and temperatures wax and wane, near-vent rates of thermomechanical deformation are likely large enough to substantially affect permeability (Germanovich and Lowell, 1992). On a longer time scale, plate movement away from the MOR itself (another mode of deformation) will also influence the overall pattern of fluid circulation.

Although we can recognize the probable importance of each of these couplings we still, almost invariably, neglect some of them in our analyses. There is as yet no quantitative model that can fully describe MOR hydrothermal circulation, or other complex, transient systems. We typically account, at most, for one or two of the couplings in each analysis, hoping to capture the essence of the system. As was the case when the first edition of this book was completed, we find that efforts to describe more than one coupling at a time (e.g., fluid flow–solute transport–heat transport) are still in their infancy.

Preface xvii

Scope of this book

The scope of this second edition of Groundwater in Geologic Processes is substantially expanded relative to the first edition. We attempt a reasonably complete introduction to the coupling of groundwater flow and deformation (Chapter 2), and have greatly expanded the treatment of compaction and diagenesis (Chapter 11), metamorphism (Chapter 12), and subsea hydrogeology (Chapter 13). However, notable omissions remain. For instance, one might reasonably expect a book by this name to cover the topic of groundwater geomorphology. Instead, we consciously avoid this important topic, other than a brief mention of the possible role of thermal pressurization in volcanic-edifice collapse (Section 8.2.3). This omission reflects a desire to limit the book to an attractive length as well as the limited geomorphic expertise of this group of authors. Instead, we refer the interested reader to other references (Higgins and Coates, 1990; Iverson and Reid, 1992; Reid and Iverson, 1992; Iverson, 1997). Karst geomorphology, in particular, is thoroughly treated elsewhere (Higgins and Coates, 1990; American Geological Institute, 2001; Culver and White, 2004).

Several reviewers of the first edition suggested more in-depth treatment of numerical-modeling techniques. We discussed this possibility among ourselves and eventually rejected it; our sense is that solution techniques are constantly evolving and therefore relatively ephemeral, in contrast to the more fundamental governing equations that we choose to emphasize here. Finally, many European reviewers of the first edition commented (justifiably) on our "North American" bias, for instance in our selection of case studies. We recognize that bias and have made some attempt to correct it in this edition. However, to a large extent it is an inevitable consequence of our own research experience and assignments as members of the US Geological Survey.

Acknowledgments

The notion of a book on this topic was originally inspired by a course that we developed for colleagues at the US Geological Survey, and several of the chapters were refined on the basis of short courses held at the Survey's National Training Center and elsewhere. We extend our gratitude to our co-instructor Evelyn Roeloffs, to the many course participants, and to numerous colleague reviewers. The following people reviewed one or more chapters of the first or second edition; unless otherwise indicated, they are members of the USGS: Jay Ague (Yale University), Barbara Bekins, Jim Bischoff, Lizet Christiansen, Greg Dipple (University of British Columbia), Hedeff Essaid, Andy Fisher (University of California-Santa Cruz), Lisl Gaal (University of Minnesota), Chris Green, Dan Hayba, Paul Hsieh, Shaul Hurwitz, Blair Jones, Ken Kipp, Lenny Konikow, Henk Kooi (Vrije Universiteit), Thomas Keijzer (Tauw BV Netherlands), Robert Lowell (Georgia Institute of Technology), Les Magoon, Michael Manga (University of California-Berkeley), Craig Manning (UCLA), Brian McPherson (New Mexico Institute of Technology), Alan Moench, Casey Moore (University of California-Santa Cruz), Mark Person (University of Indiana), Carl Renshaw (Dartmouth College), Evelyn Roeloffs, Elizabeth Rowan, Martha Scholl, Craig Simmons (Flinders University), Phil Stauffer (Los Alamos National Laboratory), Claire Tiedeman, Tomochika Tokunaga (University of Tokyo), Takuya Urakoshi (University of Tokyo), Herb Wang (University of Wisconsin) Fiona Whitaker (University of Bristol), Art White, Warren Wood, and Mary Lou Zoback. Classes taught by Paul Hsieh (at Stanford University) and Roy Haggerty (at Oregon State University) tested the first edition of this book in manuscript form; they found and corrected a number of errors in the presentation. Further necessary corrections were later identified through courses taught by Michael Manga (then at the University of Oregon) and Brian McPherson (New Mexico Institute of Technology). We also gratefully acknowledge the influence of John Bredehoeft, who first stimulated our interest in the geological aspects of hydrogeology.

Finally, we owe a debt of thanks to the USGS itself for allowing us to invest time in this effort. In our view the USGS has a strong

xviii

Acknowledgments xix

organizational interest in the topic of groundwater in geologic processes. This view is borne out to some extent by examination of the chapter titles, several of which (e.g., ore deposits, hydrocarbons, earthquakes) directly correspond to important USGS programs. Others that show no direct correspondence (e.g., geothermal processes) nevertheless represent topical areas that are significant to one or more USGS programs.

Symbols

Listed are common symbols in alphabetical order, first in the Latin, then the Greek alphabets. Other, less frequently used symbols are defined where they appear in the text. Dimensions given in square brackets are M, mass, L, length, t, time, E, energy, T, temperature, and C, electrical charge.

 $A = \operatorname{area} \left[L^2 \right]$

- A_{o} = radiogenic heat production measured near the Earth's surface [E/L³-t]
- b = thickness[L]
- B = three-dimensional loading efficiency [dimensionless]
- c = isochoric heat capacity [E/m-T]
- $c_{\rm b}$ = bulk compressibility [Lt²/M]
- $c_{\rm f}$ = bulk compressibility of fluid [Lt²/M]
- c_p = bulk compressibility for plastic deformation [Lt²/M]
- $c_{\rm s}$ = bulk compressibility of nonporous solids [Lt²/M]
- c_u = uniaxial (confined) compressibility $[(1 + \nu)/3(1 \nu)]c_b$ [Lt²/M]
- c'_{u} = effective or long-term uniaxial (confined) compressibility [Lt²/M]
- C = aqueous concentration [M/L³]
- C_d = dimensionless concentration [dimensionless]
- C_{eq} = concentration at equilibrium with respect to a given mineral [M/L³]
- C_i = concentration in inflowing groundwater [M/L³]
- $C_{\rm L}$ = uniaxial loading coefficient $[(1 n)(\rho_{\rm s} \rho_{\rm f})]/\rho_{\rm f}$ [dimensionless]
- $C_{\rm o}$ = concentration in outflowing groundwater [M/L³]
- $C_{\rm p}$ = concentration in precipitation [M/L³]
- $C_{\rm s}$ = sorbed concentration (Chapter 3) [M/M]
 - = solute concentration in inflowing surface water (Chapter 10) [M/L³]
- C_{o} = reference concentration or concentration at an initial time [M/L³]

List of symbols xxi

CEC	= the exchange capacity of a porous medium [M/M]
d	= depth [L]
$d_{\rm m}$	= mean or median grain diameter (particle size) [L]
D	= hydrodynamic dispersion (Chapters 3, 5) $[L^2/t]$
	= hydraulic diffusivity (Chapter 8) $[L^2/t]$
D'_{ij}	= component of diffusion coefficient matrix for a
5	multicomponent system [L ² /t]
$D_{\rm k}$	= Damkohler number [dimensionless]
D_{L}	= coefficient of longitudinal dispersion $[L^2/t]$
$D_{\rm m}$	= coefficient of molecular diffusion in a porous medium $[L^2/t]$
$D'_{ m m}$	= coefficient of molecular diffusion per unit area of effective
	porosity [L ² /t]
D_{T}	= coefficient of transverse dispersion $[L^2/t]$
D_{w}	= coefficient of molecular diffusion in open water $[L^2/t]$
е	= activation energy [E/M]
Ε	= Young's modulus for deformation (Chapter 2)[M/L -t ²]
Ε	= impelling force $[ML/t^2]$
E_j	= gradient in the electrochemical potential of the <i>j</i> th ion [E/M-L]
F	= Faraday's constant $[C/M]$
g	= gravitational acceleration $[L/t^2]$
G	= shear modulus, $E/2(1 - \nu)$ [M/L-t ²]
h	= hydraulic head [L]
Н	= enthalpy [E]
k	= intrinsic permeability [L ²]
k_i	= kinetic rate constant for a given mineral i (i = A, B, C)
	[variable]
<i>k</i> _r	= relative permeability [dimensionless]
Κ	= hydraulic conductivity [L/t]
K _{ac}	= equilibrium constant for an aqueous complex [variable]
K _d	= linear sorption distribution coefficient $[L^3/M]$
K _{eq}	
K_i	
K _m	
$K_{\rm r}$	= a reaction-rate constant [variable]
$l_{\rm s}$	= elevation of the land surface [L]
L	= characteristic length or distance [L]
т	= empirical exponent in a kinetic rate equation [dimensionless]
$m_{\rm s}$	= the molality of a species [M/M]
М	= a point source of mass [M]
п	= porosity $[L^3/L^3$, dimensionless]
n _e	= effective porosity [dimensionless]
n _{ir}	= irreducible porosity [dimensionless]
n _s	= porosity at ground surface [dimensionless]

xxii List of symbols

Nu	= Nusselt number [dimensionless]
P	= pressure [M/L-t ²]
Pe	= Peclet number [dimensionless]
q	= volumetric flow rate per unit area (volume flux, specific
4	discharge or Darcy velocity) [L/t]
q_{a}	= advective flux of a solute $[M/L^2-t]$
$q_{\rm d}$	= diffusive flux of a solute $[M/L^2-t]$
$q_{ m h}$	= conductive heat flux per unit area $[E/L^2-t]$
$q_{ m h}^{ m m}$	= reduced conductive heat flux per unit area $[E/L^2 - t]$
$q_h q_i$	= diffusive flux of the <i>i</i> th ion $[M/L^2-t]$
$\overset{q_l}{Q}$	= total volumetric flow rate $[L^3/t]$
\tilde{Q}_{e}	= volumetric flow rate of evaporating basin water $[L^3/t]$
\hat{Q}_{i}	= volumetric flow rate of inflowing groundwater $[L^3/t]$
\hat{Q}_{0}	= volumetric flow rate of outflowing groundwater $[L^3/t]$
$\hat{Q}_{\rm p}$	= volumetric flow rate of precipitation $[L^3/t]$
\hat{Q}_{s}^{p}	= source or sink of solute (Chapter 3) $[M/L^3-t]$
2.5	= volumetric flow rate of inflowing surface water (Chapter
	10) $[L^3/t]$
r	= radial distance (Chapter 1) or a radius of curvature (Chapter
	7) [L]
R	= general source/sink term for mass, chemical reactions, or
	heat [variable]
	= gas constant (Chapter 7) [E/M-mol-T]
	= D/H or ${}^{18}O/{}^{16}O$ ratio (Chapter 12) [dimensionless]
$R_{\rm c}$	= rate of calcite dissolution [variable]
R_{f}	= retardation factor [dimensionless]
$R_{\rm p}$	= rate of porosity development $[L^3/L^3-t]$
Ra	= Rayleigh number [dimensionless]
S	= storage coefficient (storativity) [dimensionless]
Ss	= specific storage [1/L]
s'_S	= specific storage (compressible solids) [1/L]
s _{S3}	= three-dimensional specific storage [1/L]
s'_{S3}	= modified three-dimensional specific storage [1/L]
<i>s</i> ₀	= specific surface $[L^2/L^3]$
S	= volumetric saturation $[L^3/L^3, dimensionless]$
$S_{ m y}$	= specific yield [dimensionless]
t	= time [t]
t _d	= dimensionless time [dimensionless]
Т	= temperature [T]
	= transmissivity (Chapters 1 and 9) $[L^2/t]$
Τ	= surface traction [M/t ² L]
и	= displacement vector (u, v, w) [L]

List of symbols xxiii

u, v, w	= displacement in the x, y, z coordinate directions [L]
υ	= average linear velocity (seepage velocity) $[L/t]$
V	= volume [L ³]
V _e	= evaporated basin volume (Chapter 10) [dimensionless]
$V_{\rm f}$	= volume of fluid $[L^3]$
V _p	= volume of pores $[L^3]$ = mass fraction steam (Chapter 8) or H ₂ O or CO ₂ in an
x	= mass fraction steam (Chapter 8) of H_2O of CO_2 in an H_2O – CO_2 mixture (Chapter 12) [dimensionless]
$x_{\rm d}$	= dimensionless distance in the x direction
Xa X	= Lagrangian x coordinate [L]
X'	= translational transformation variable [L]
Xa	= exchange site for ion a on a mineral surface [dimensionless]
Ŷ	= Lagrangian y coordinate [L]
Ζ	= elevation above a datum, elevation head, or vertical
	Cartesian coordinate [L]
Z_i	= ionic charge of the <i>i</i> th ion [dimensionless]
Ζ	= Lagrangian z coordinate [L]
	coefficient of the model and and (Charters A and 12)
α	= coefficient of thermal expansion (Chapters 4 and 12) [Δ L/L-T, 1/T]
	= effective stress coefficient, $1 - (c_s/c_b)$ (Chapter 2)
	[dimensionless]
	= contact angle (Chapter 7)
$\alpha_{\rm L}$	= longitudinal dispersivity [L]
$\alpha_{\rm T}$	= transverse dispersivity [L]
	= porous medium linear thermal expansivity (Chapter 2)
	[1/T]
α_{Tb}	= porous medium bulk thermal expansivity $[1/T]$
α_{Tf}	= fluid thermal expansivity $[1/T]$
α_{Tp}	= pore thermal expansivity [1/T]
$\alpha_{\rm Ts}$	= mineral or solid thermal expansivity $[1/T]$
β	= fluid compressibility $[Lt^2/M]$
δ	= dirac delta function (Chapter 3) [dimensionless] $\begin{bmatrix} (B_{1}, \dots, (B_{n})) & 1 \end{bmatrix} \mapsto 1000 (Chapter 12)$
ΛC_{0}	= $[(R_{\text{sample}} / R_{\text{SMOW}}) - 1] \times 1000$ (Chapter 12) = mass of dissolved calcite required to reach equilibrium
ΔCa_s	$[M/L^3]$
Δx	= discretization length in the x direction for a computational
	grid [L]
ε	= strain [$\Delta L/L$, dimensionless]
ε_{ij}	= strain component; $i = j$ indicates contraction/extension;
2	$i \neq j$ indicates shear distortion [dimensionless]
$\varepsilon_{\rm kk}$	= volumetric strain, $\varepsilon_{xx} + \varepsilon_{yy} + \varepsilon_{zz} [\Delta L^3/L^3, dimensionless]$

CAMBRIDGE

xxiv List of symbols

${\cal E}_{ m v}$	= vertical strain [$\Delta L/L$, dimensionless]
ϕ	= electrostatic potential (Chapter 3) $[E/C]$
arphi	= fluid potential per unit mass (Chapter 8) [E/M]
	= angle of internal friction (Chapter 9); $\tan \phi$ is also
	known as the <i>coefficient of friction</i>
γ	= change in fluid density per change in solute
	concentration [dimensionless]
$\gamma, \gamma_1, \gamma_2$	= coefficients for Athy-type compaction (Chapter 2)
	$[Lt^2/M]$
γm	= activity coefficient of aqueous species m (Chapter 3)
	[dimensionless]
Г	= a volumetric source or sink of fluid or equivalent rate of
	porosity loss or gain (Chapter 5) $[L^3/L^3/t]$
λ	= poroelastic coefficient, $2\alpha(1-2\nu)/3(1-\nu)$ (Chapter 2)
	[dimensionless]
	= radioactive decay constant $[1/t]$
λ'	= loss-by-diffusion constant (Chapter 5) $[1/t]$
Λ	= thermal response coefficient, $\alpha_{Tf} - \alpha_{Tp} [1/T]$
Λ'	= thermal response coefficient for uncoupled flow
	equation, $\alpha_{Tf} + (\lambda/n)\alpha_{Tb} - \alpha_{Tp}$ [1/T]
μ	= dynamic viscosity [M/L-t]
$\mu_{ m i}$	= chemical potential of the <i>i</i> th ion (Chapter 3) $[E/M]$
$\mu_{ m s}$	= dynamic viscosity for deformation of porous medium
	[M/L-t]
v	= Poisson's ratio (Chapter 2) [dimensionless]
	= stoichiometric coefficient in a chemical reaction
	[dimensionless]
$\pi_{ m h}$	= osmotic head [L]
ρ	$= \text{density} [M/L^3]$
$ ho_{ m f}$	= density of fresh water (Chapter 10) $[M/L^3]$
$ ho_{0}$	= reference fluid density $[M/L^3]$
ρ_{s}	= density of salt water (Chapter 10) $[M/L^3]$
θ	= angle between failure plane and greatest principal stress
σ	= general designation of stress, positive for compression;
	one of three mutually orthogonal principal stresses
	[M/L-t ²]
	= membrane efficiency (Chapters 1 and 3) [dimensionless]
-	= surface tension (Chapter 7) $[M/t^2]$
$\sigma_{\rm d}$	= differential stress, $\sigma_1 - \sigma_3$ [M/L-t ²]
$\sigma_{ m e}$	= effective normal stress on an arbitrarily oriented plane, $\sigma = \frac{P}{r} \frac{[M/L] + r^2}{r^2}$
-	$\sigma_{\rm n} - P_{\rm f} \left[{\rm M/L-t^2} \right]$
σ_{ij}	= stress component in i, j, k coordinate system; i is direction permulte surface sated on i in direction of
	direction normal to surface acted on; j is direction of

CAMBRIDGE

Cambridge University Press 0521603218 - Groundwater in Geologic Processes, Second Edition Steven E. Ingebritsen, Ward E. Sanford and Christopher E. Neuzil Frontmatter More information

List of symbols xxv

	stress; $i = j$ indicates normal stress; $i \neq j$ indicates	
	shear stress [M/L-t ²]	
σ_{kk}	= sum of normal stresses, $\sigma_{xx} + \sigma_{yy} + \sigma_{zz}$ [M/L-t ²]	
$\sigma_{\rm kk}/3$	= mean normal stress [M/L-t ²]	
$\sigma_{\rm n}$	= normal stress on an arbitrarily oriented plane $[M/L-t^2]$	
$\sigma_{ m v}$	= vertical total stress $[M/t^2-L]$	
$\sigma'{}_{ m v}$	= vertical effective stress $[M/t^2-L]$	
$\sigma_1, \sigma_2, \sigma_3$	= greatest, intermediate, and least principal stress	
	$[M/L-t^2]$	
$\bar{\sigma}$	= "hydrostatic" or uniform confining stress $[M/L-t^2]$	
τ	= tortuosity (Chapter 3) [L/L]	
	= shear stress (Chapter 9) $[M/L-t^2]$	
$ au_0$	= cohesion [M/L-t ²]	
ω	= coefficient for viscous deformation of porous solid	
	[Lt/M]	
ζ	= one-dimensional loading efficiency [dimensionless]	
()' used with stress indicates <i>effective</i> rather than <i>total</i>		

(^) indicates increase or decrease in a quantity

Subscripts

Unless otherwise locally redefined, these subscripts have the following meanings. They mainly apply to the description of multiphase systems. For example, ρ_g , ρ_o , and ρ_s are the densities of gas, oil, and steam, respectively.

atm refers to atmospheric conditions

- c refers to capillary effects
- f refers to the fluid mixture in place (either a single phase or a twophase mixture)
- g refers to gas
- m refers to the porous medium
- o refers to oil
- r refers to the rock
- s refers to steam
- v refers to void space (porosity)
- w refers to liquid water
- 0 refers to an initial state (Chapter 2)